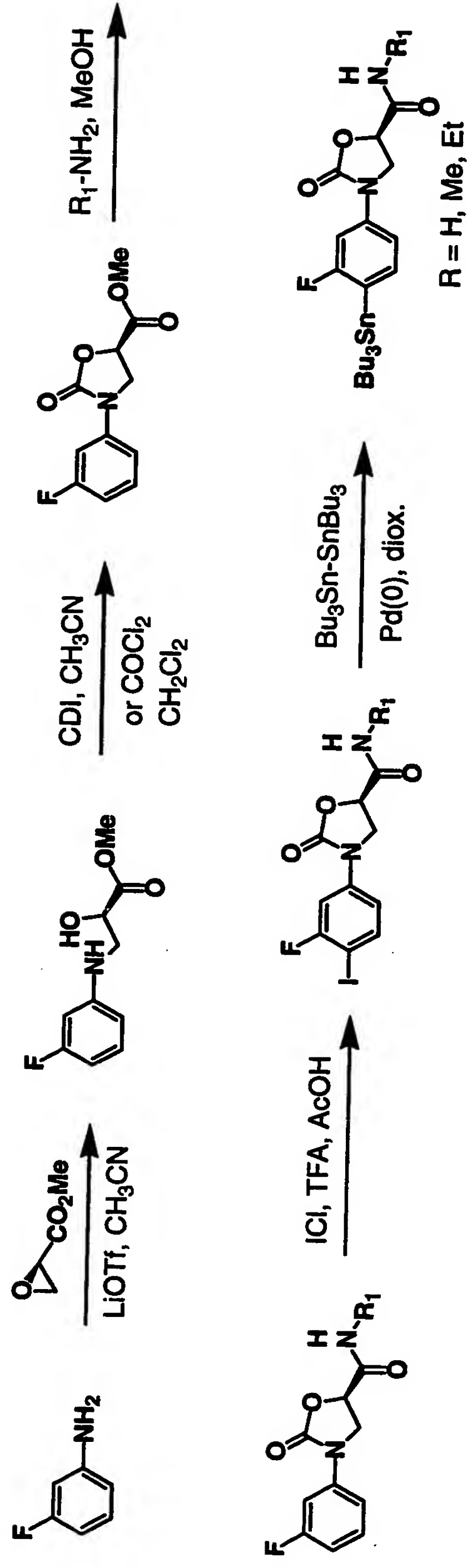
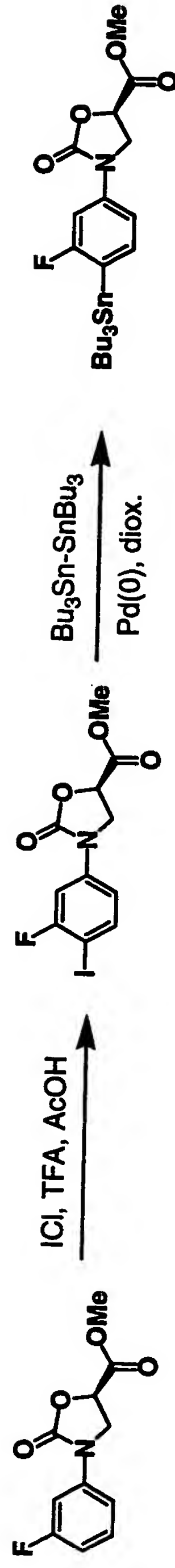


FIGURE I

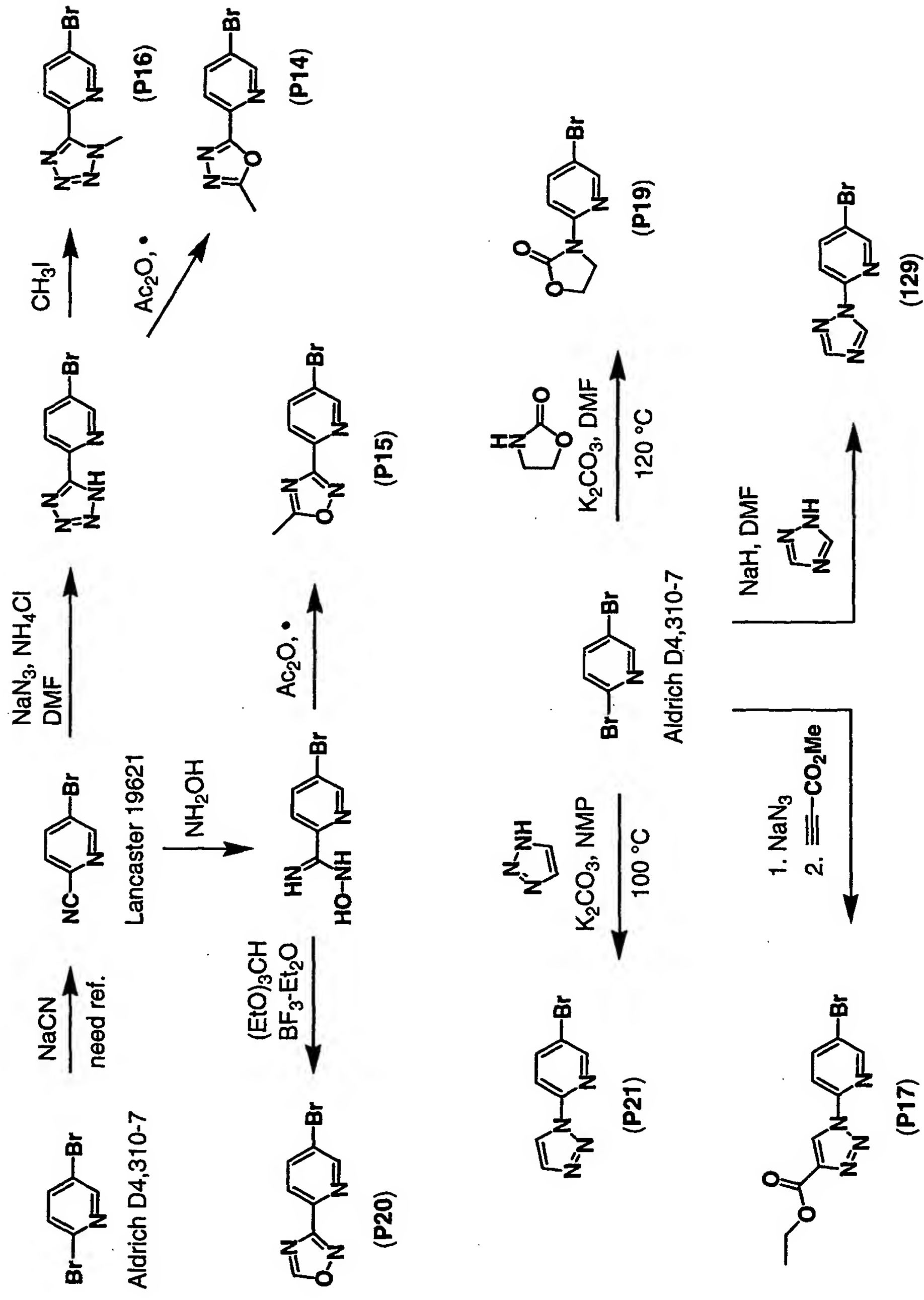
C-5 Carboxamides



C-5 Ester



Synthetic Routes to Key Heterocyclic Pyridines/Pyrimidines



Synthetic Routes to Key Heterocyclic Pyridines/Pyrimidines



FIGURE IV

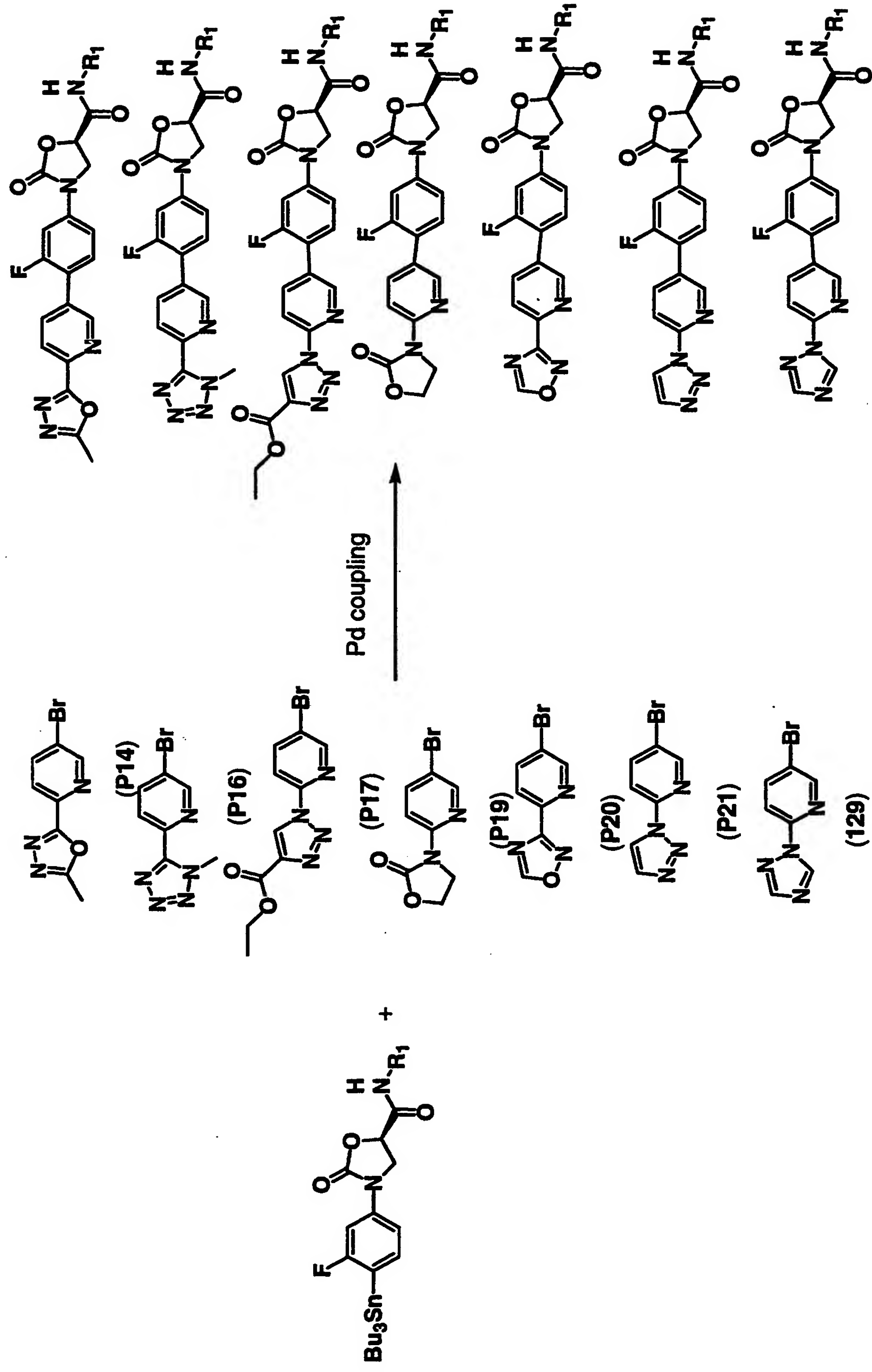


FIGURE VI

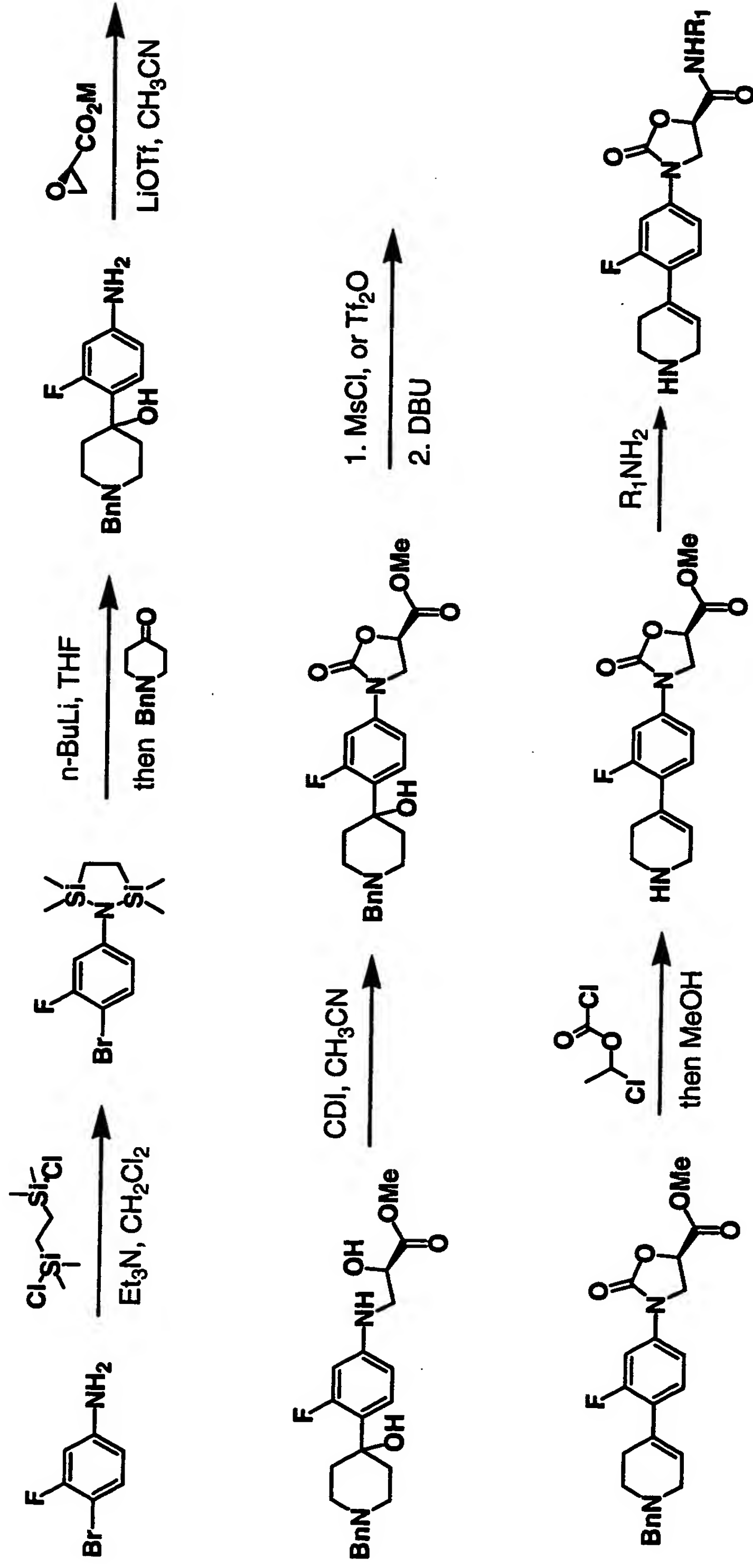


FIGURE VII

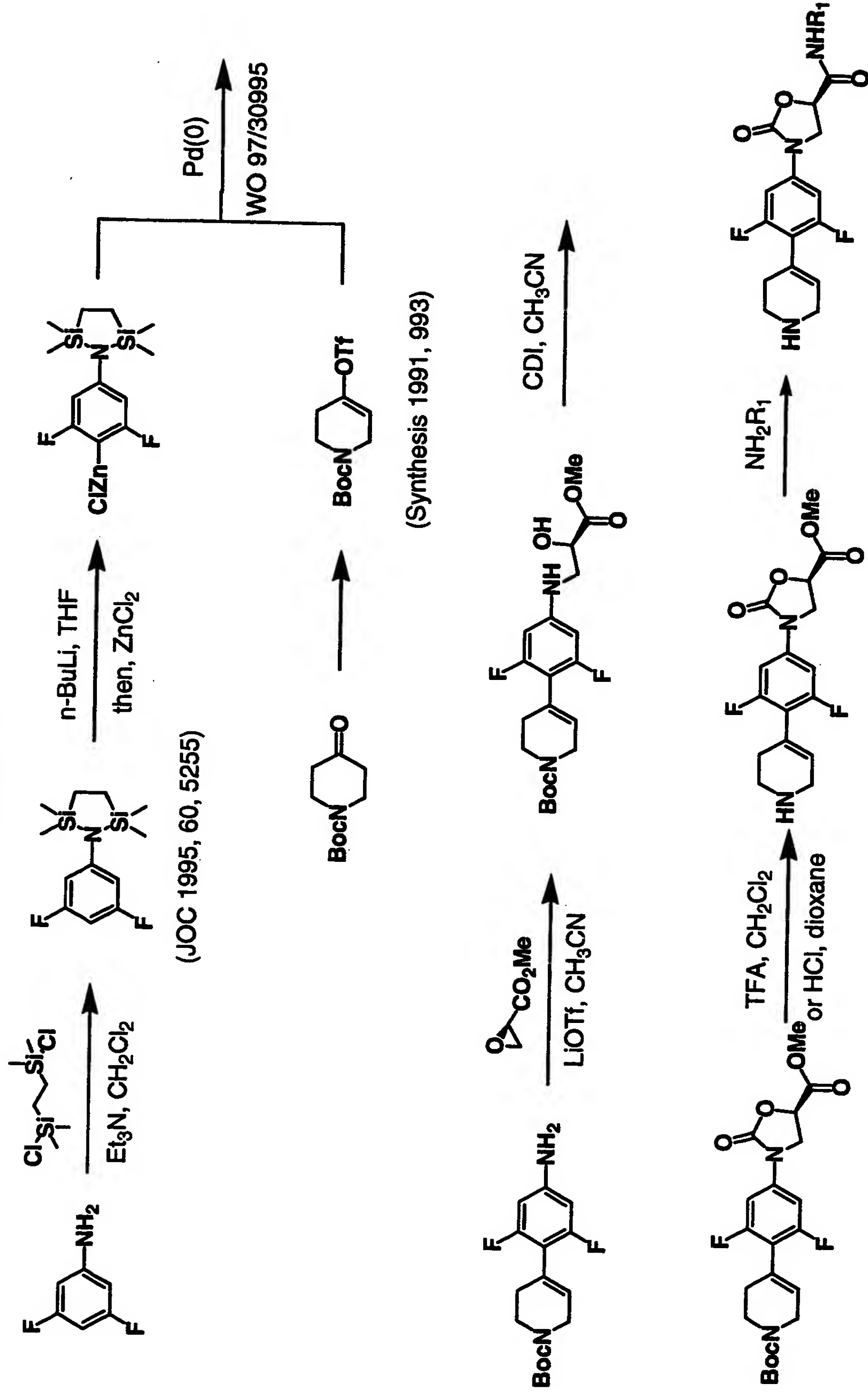
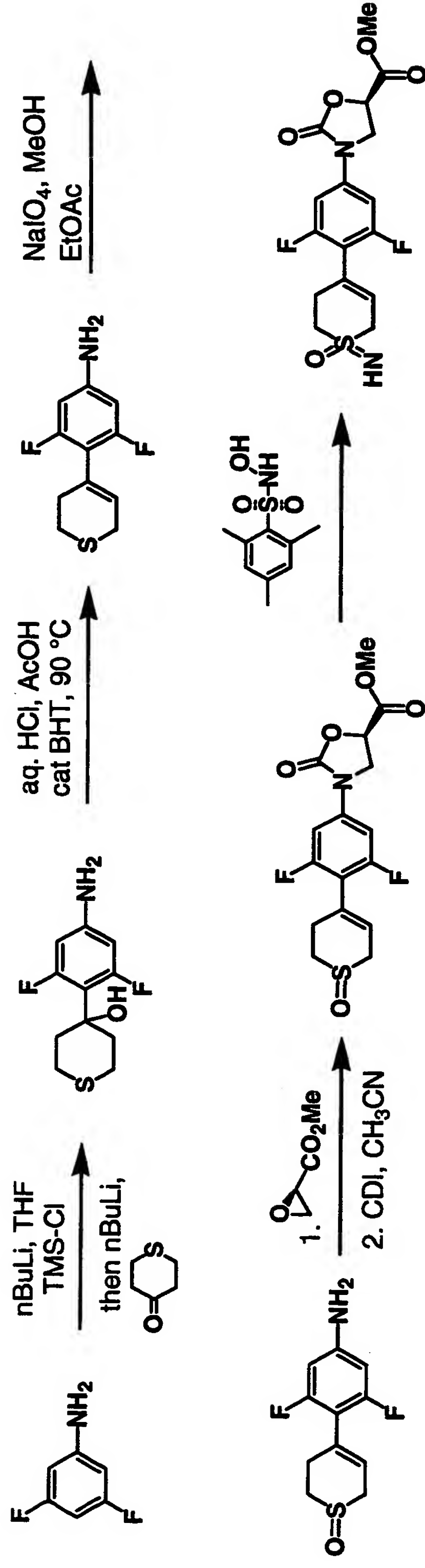
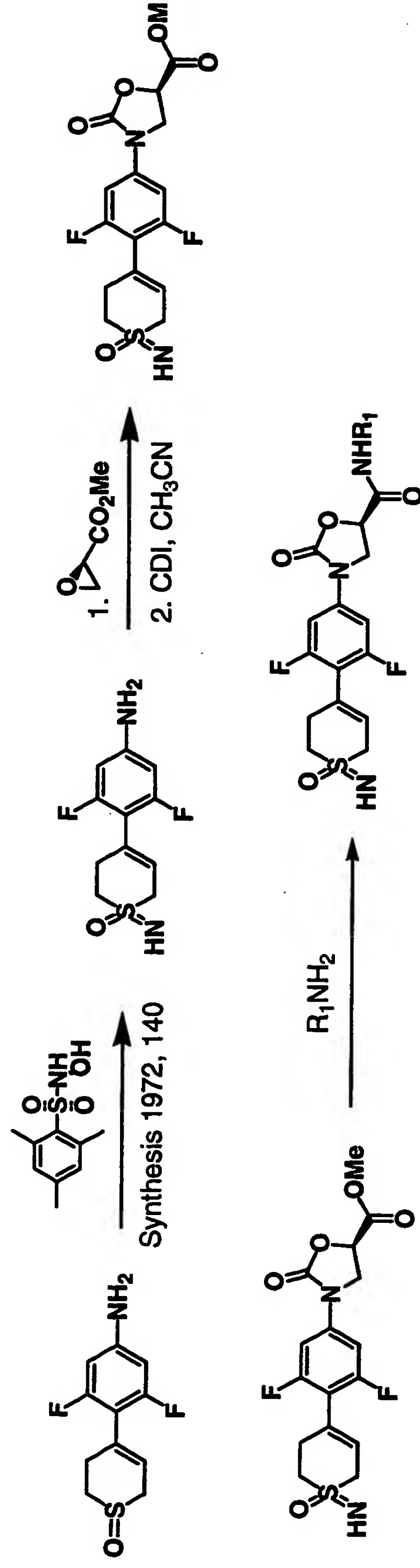


FIGURE VIII



Alternate route:



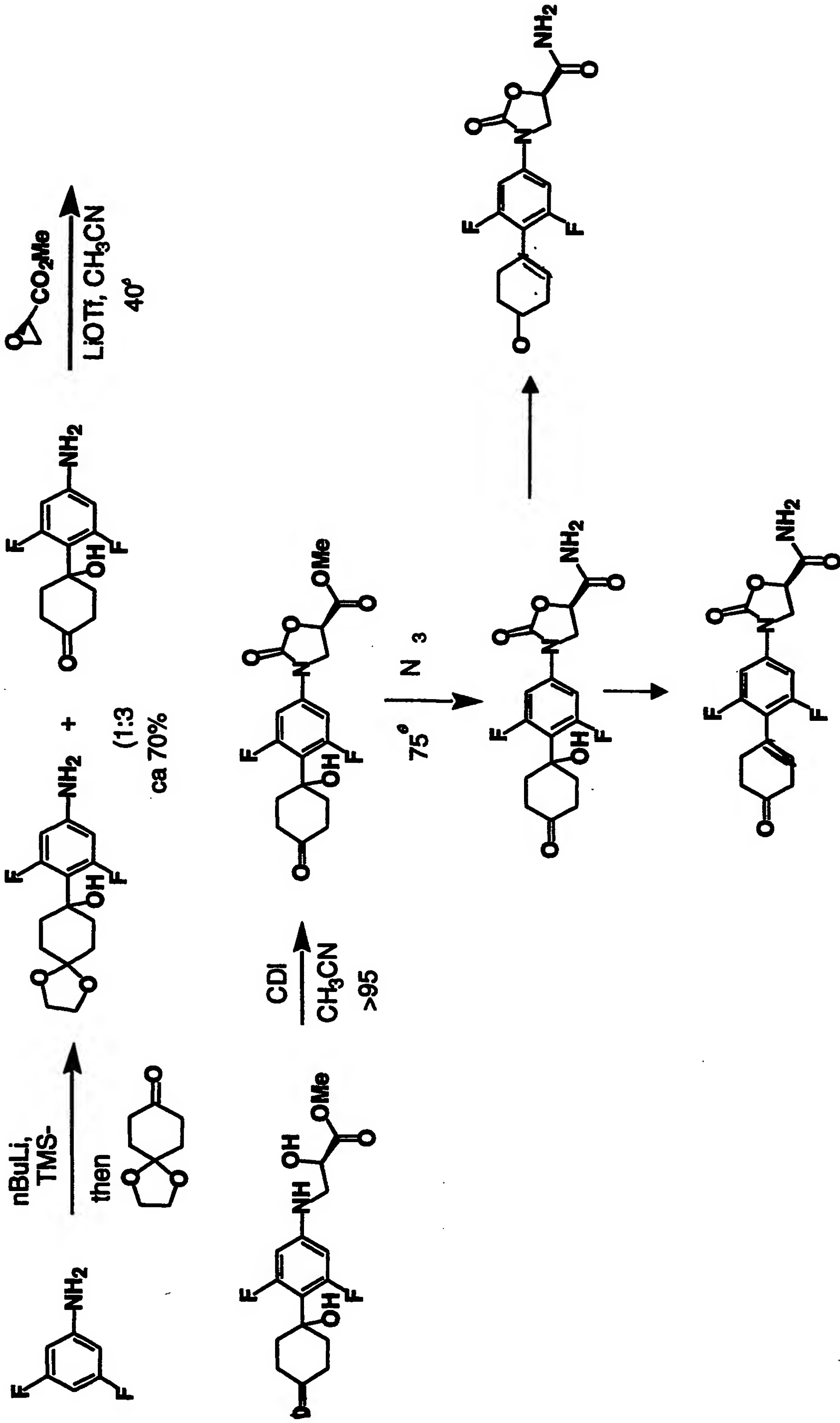
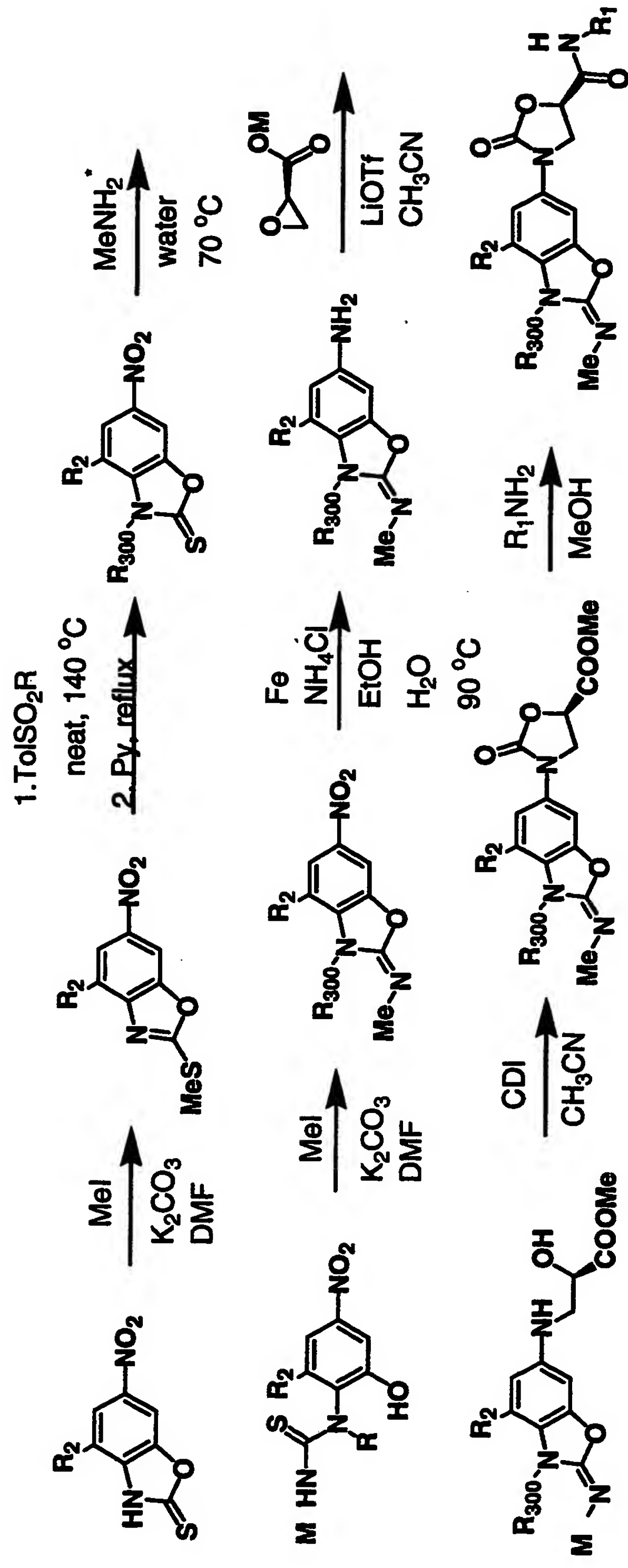


FIGURE X



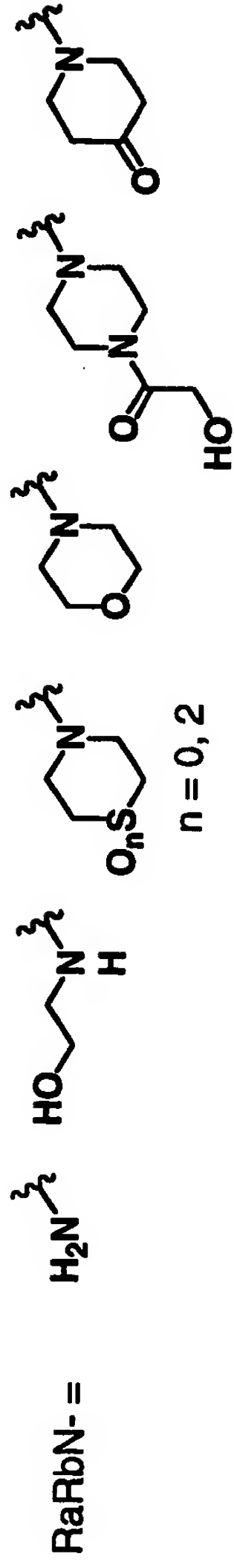


FIGURE XV

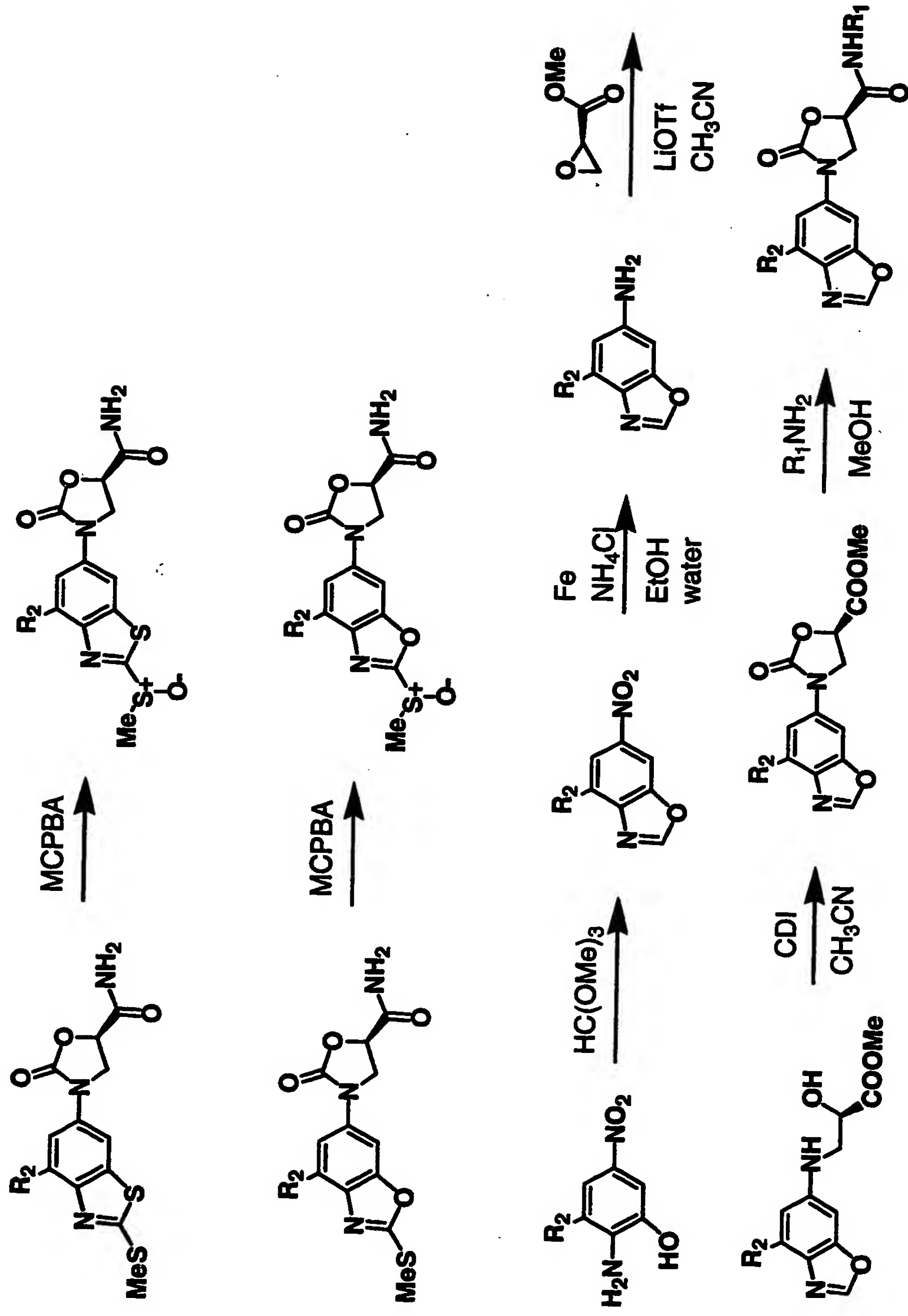


FIGURE XVII

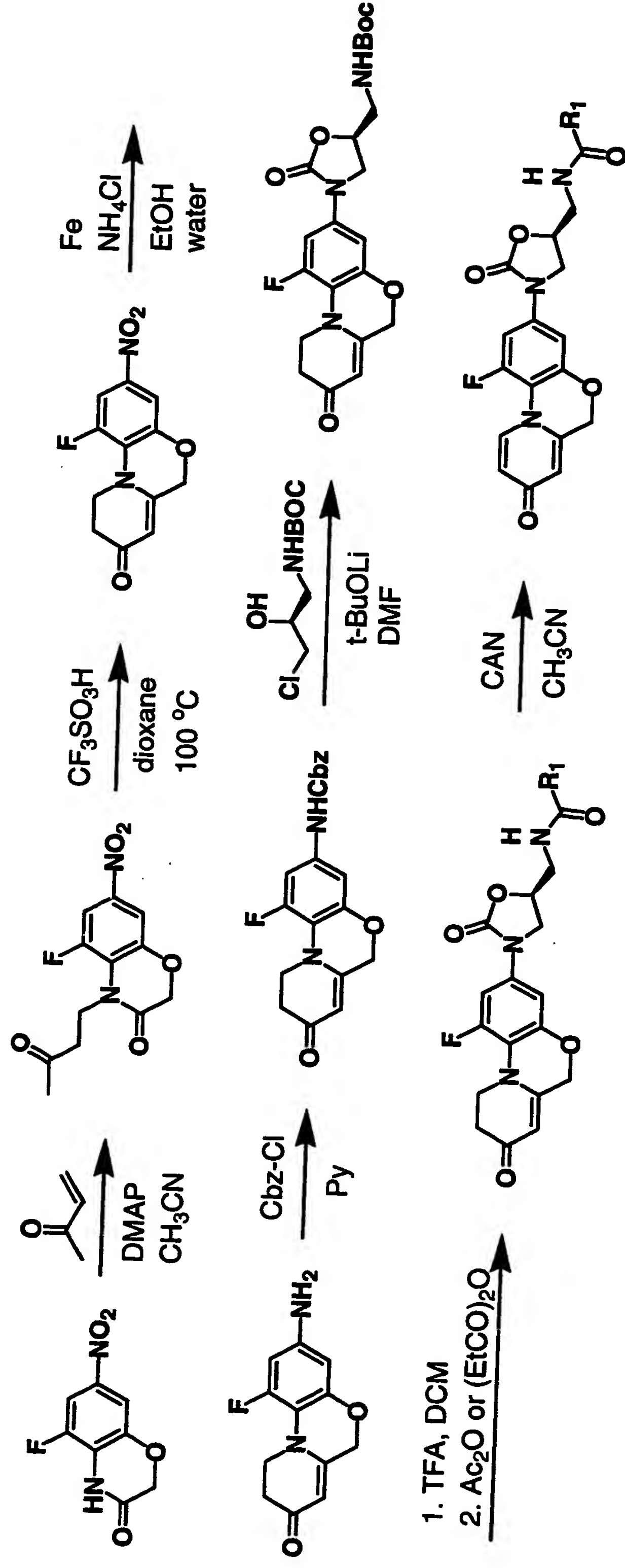


FIGURE XVIII

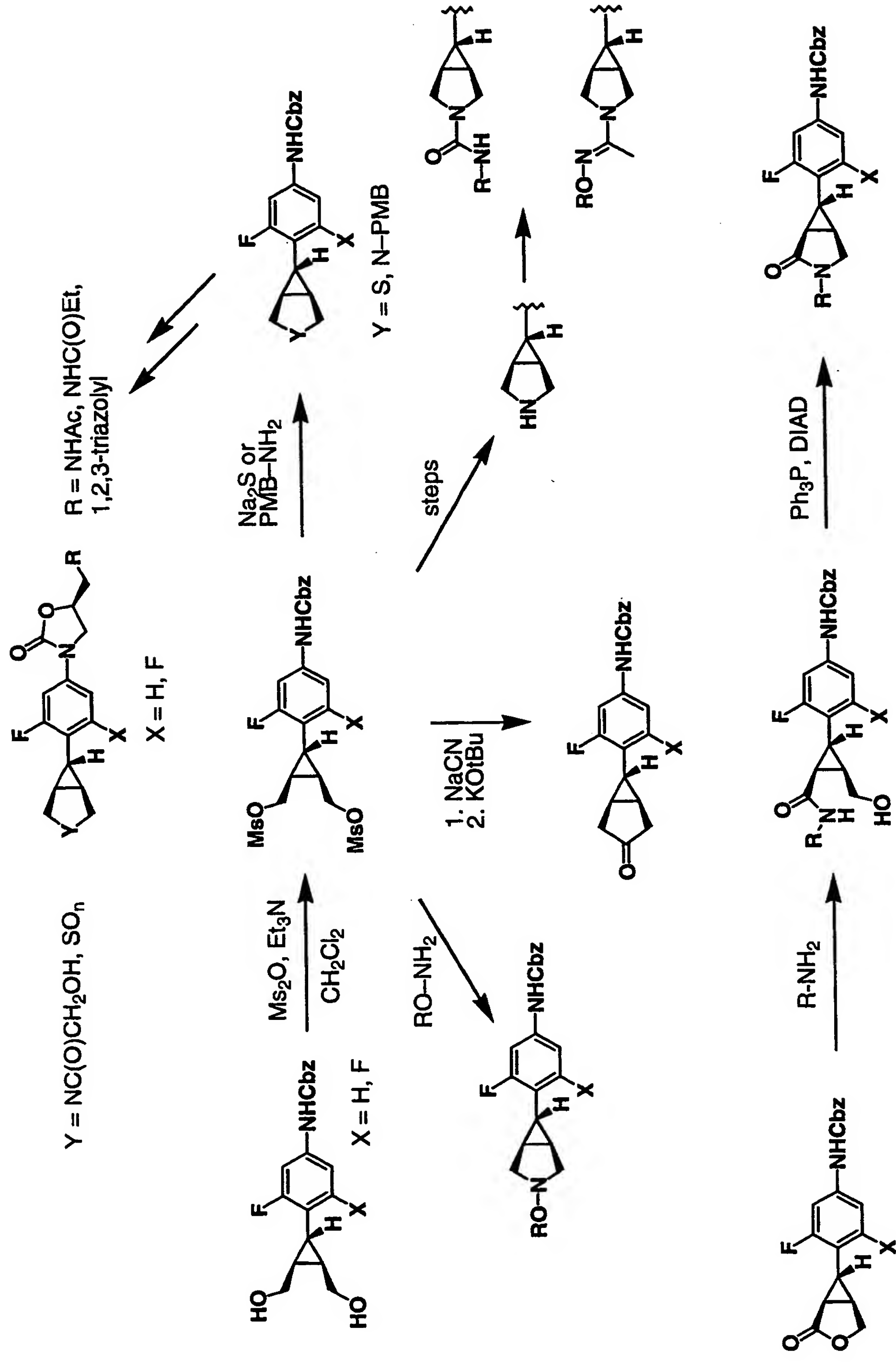


FIGURE XIX

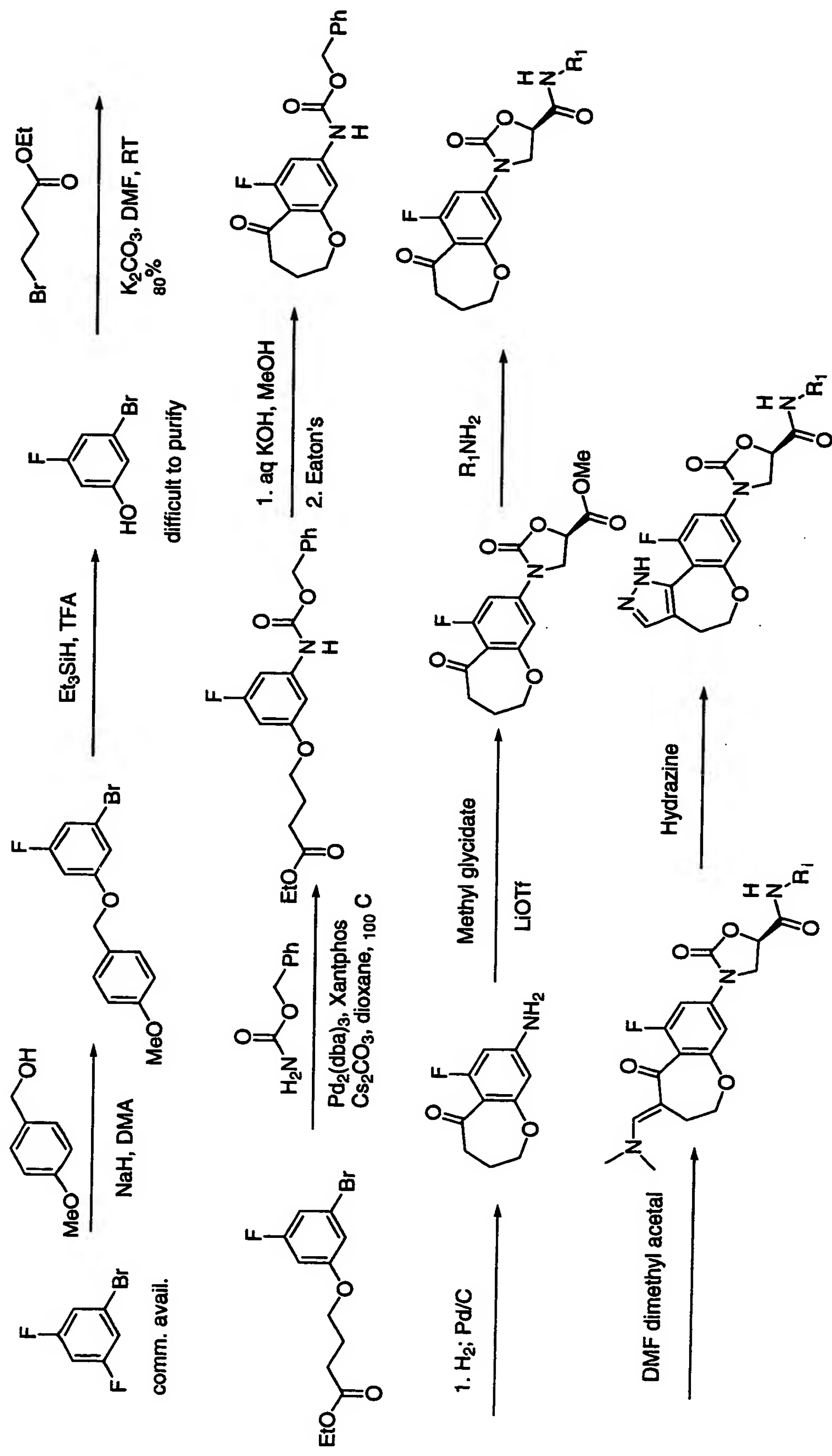


FIGURE XX

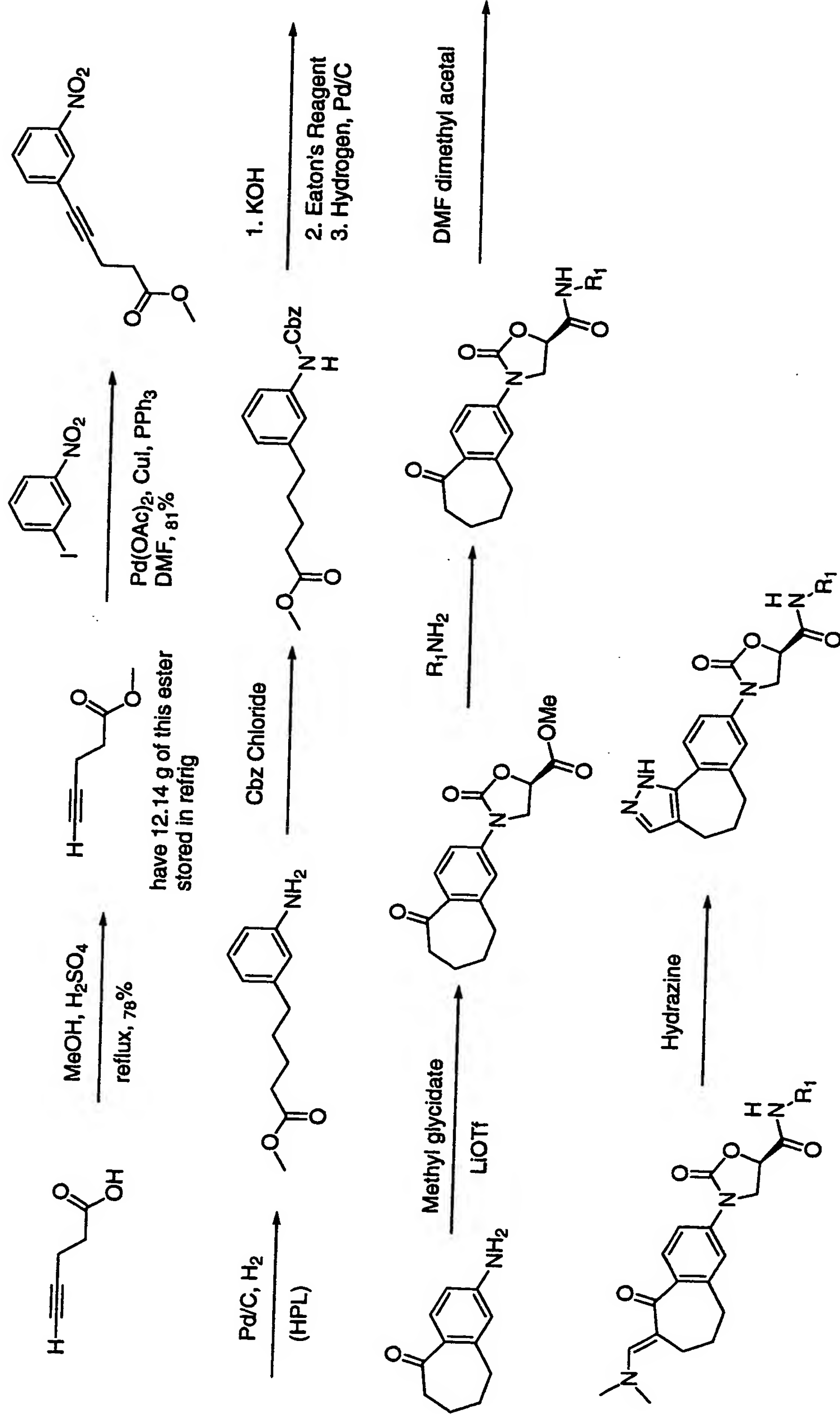


FIGURE XXI

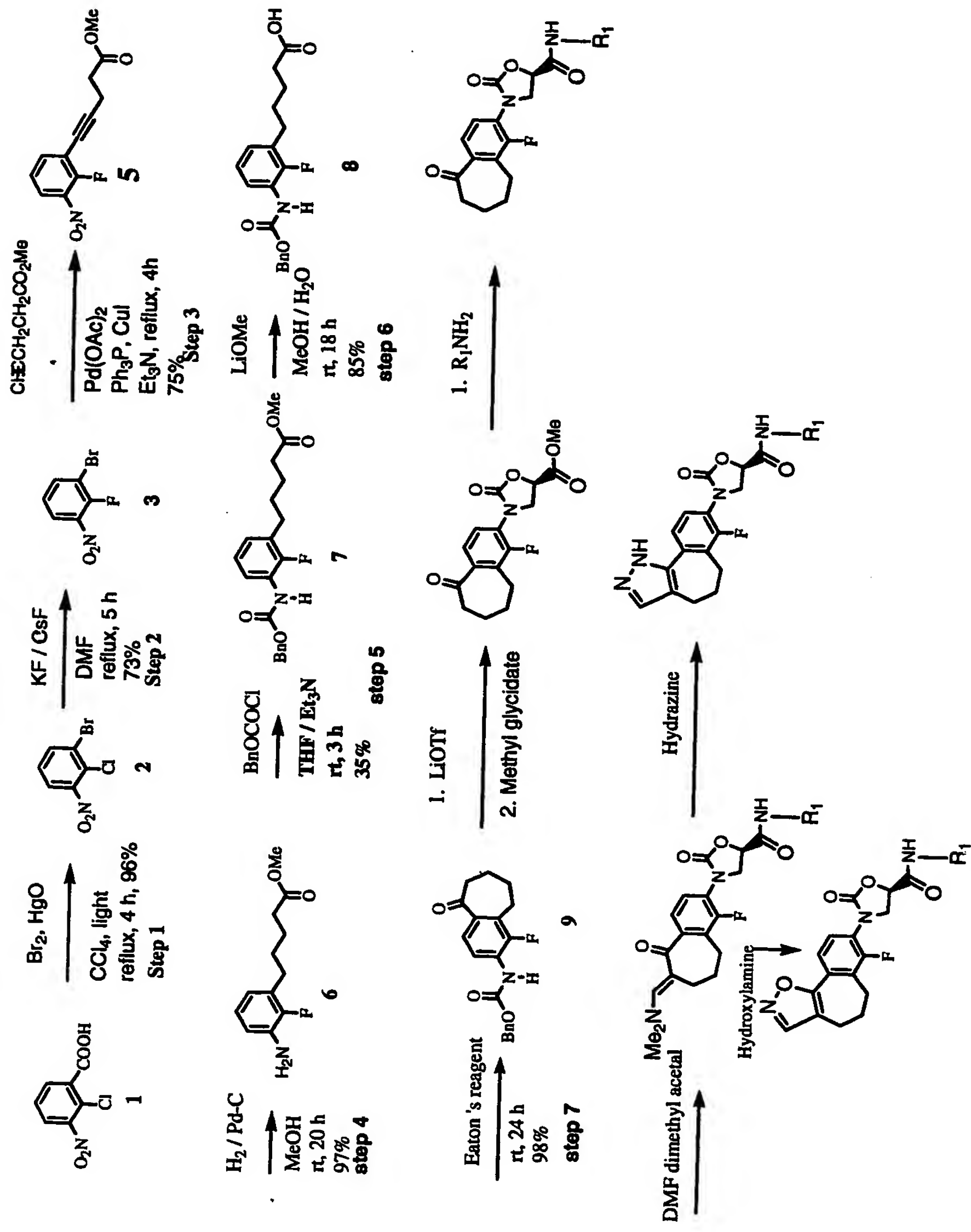


FIGURE XXII

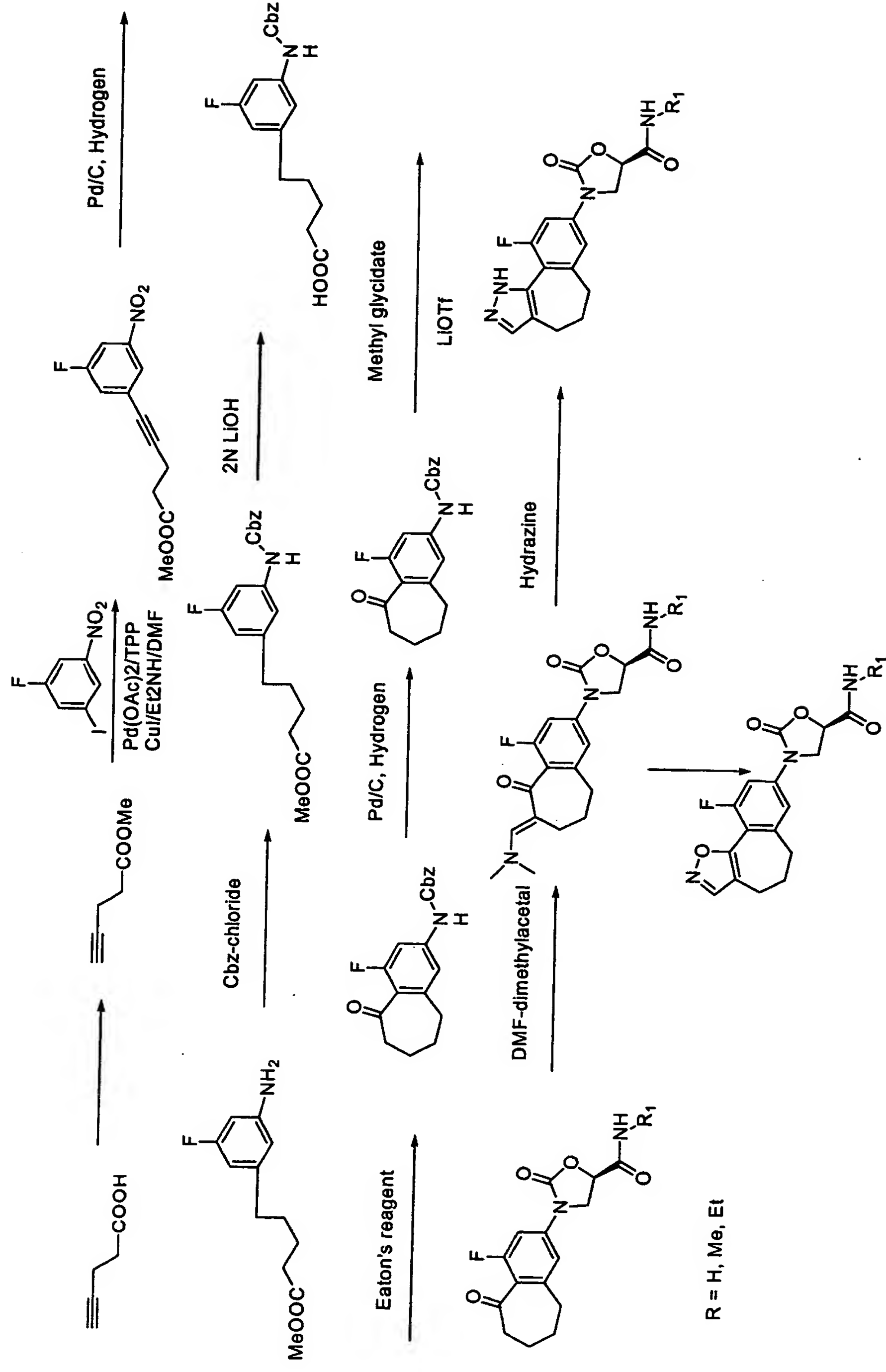
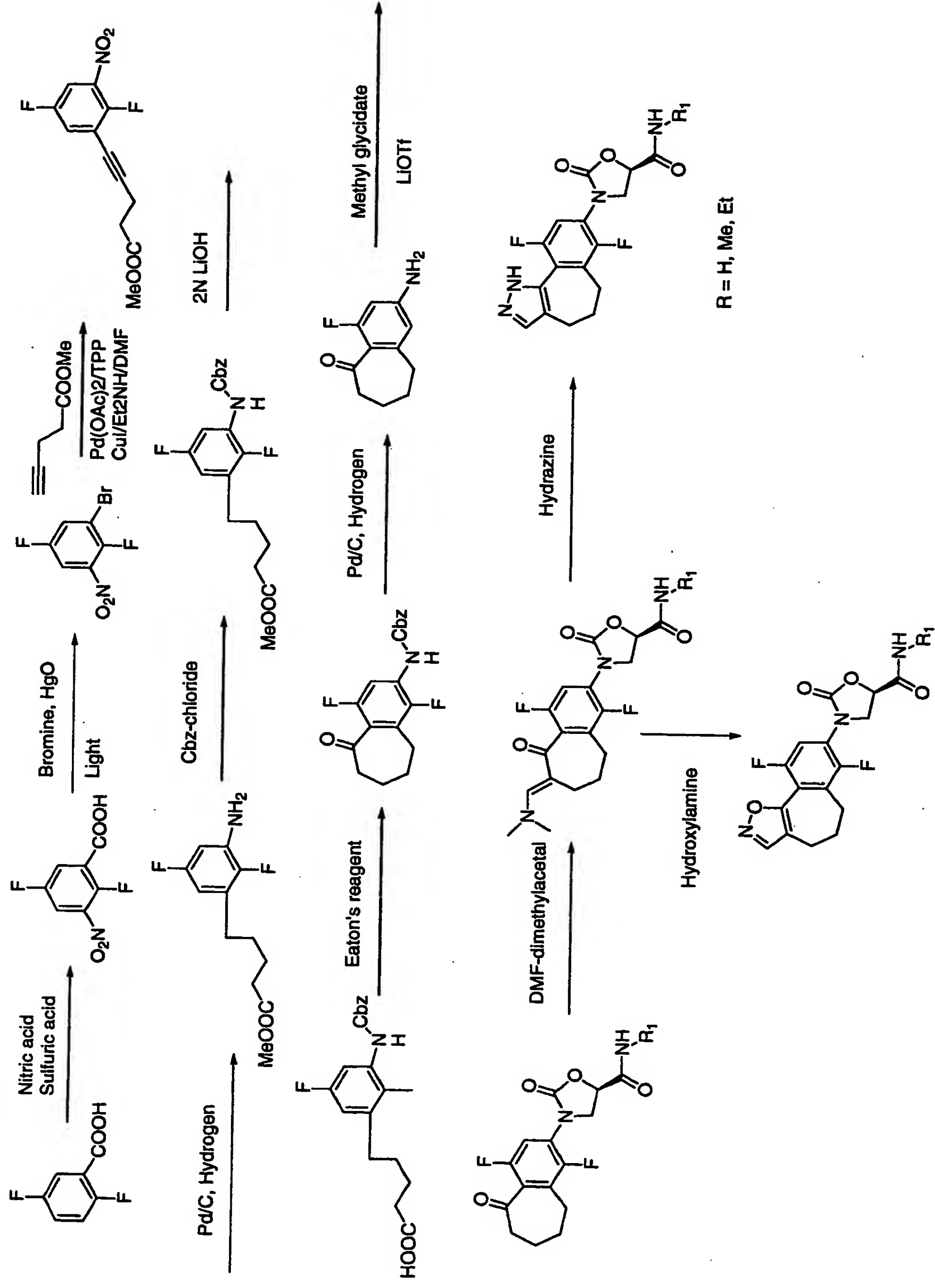


FIGURE XXIII



The reaction scheme illustrates the synthesis of 1,2,3,4-tetrahydro-8H-benzo[7,8-b]pyrido[4,3-b]indole derivatives through several steps:

- Starting Materials:** 1-bromo-3,4-difluorobenzene and methyl 3-ethynyl-3-oxopropionate.
- Step 1:** Coupling reaction using $\text{Pd}(\text{OAc})_2/\text{TPP}$ and $\text{CuI}/\text{Et}_2\text{NH}/\text{DMF}$ to form the ethynyl-substituted intermediate.
- Step 2:** Hydrogenation using Pd/C and H_2 to saturate the alkyne.
- Step 3:** Hydrolysis using 2N LiOH to yield the carboxylic acid intermediate.
- Step 4:** Activation of the carboxylic acid using 1. Oxalyl chloride and 2. Aluminum chloride to form the acid chloride.
- Step 5:** Reaction with 3,4-difluorobenzonitrile to form the nitrile-substituted intermediate.
- Step 6:** Reduction using LiOTf and methyl glycidate to form the amine-substituted intermediate.
- Step 7:** Reaction with RNH_2 to form the amine-substituted intermediate.
- Step 8:** Reaction with $\text{DMF-dimethylacetal}$ to form the amine-substituted intermediate.
- Step 9:** Reaction with Hydrazine to form the hydrazine-substituted intermediate.
- Step 10:** Reaction with Hydroxylamine to form the hydroxylamine-substituted intermediate.

FIGURE XXV

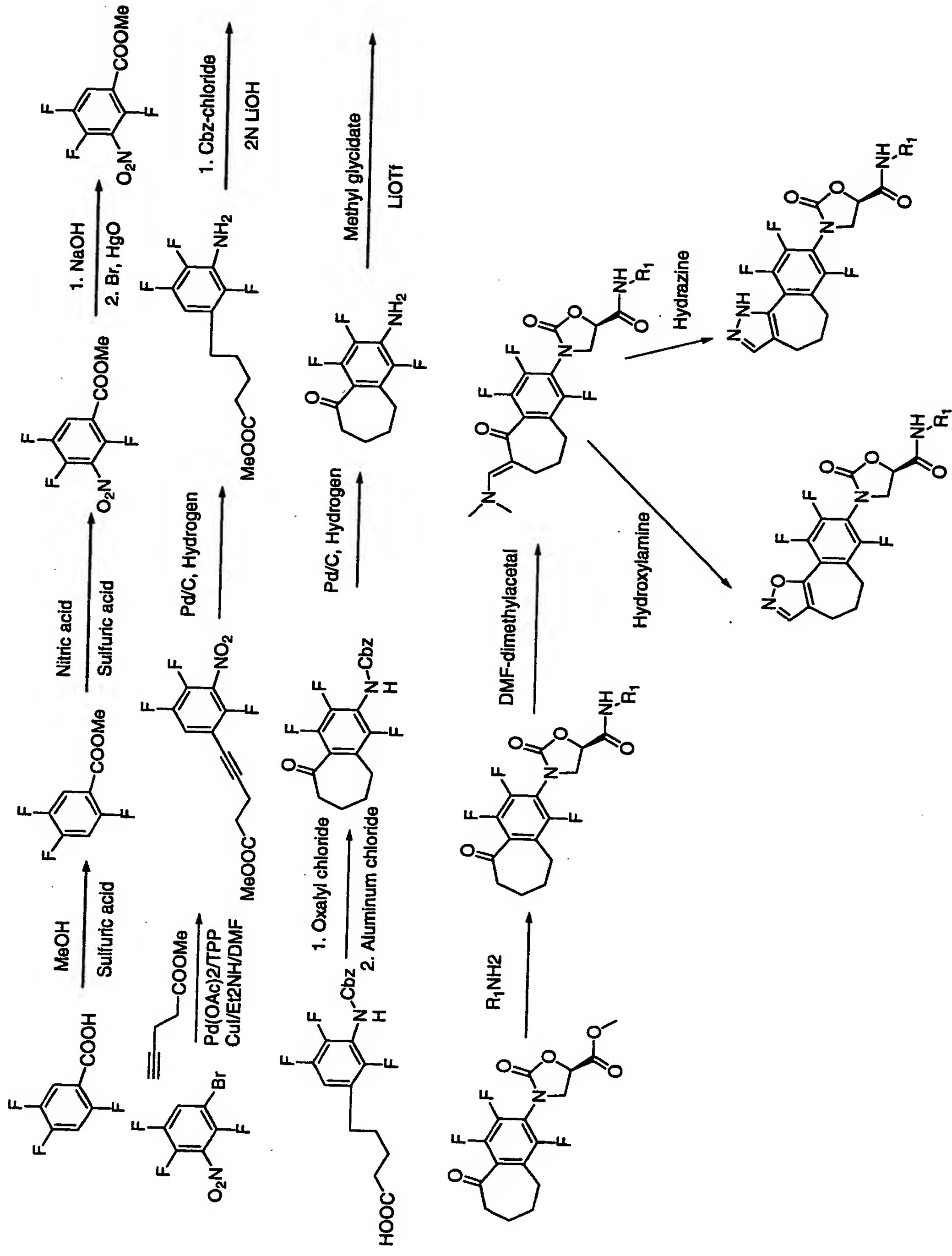


FIGURE XXVI

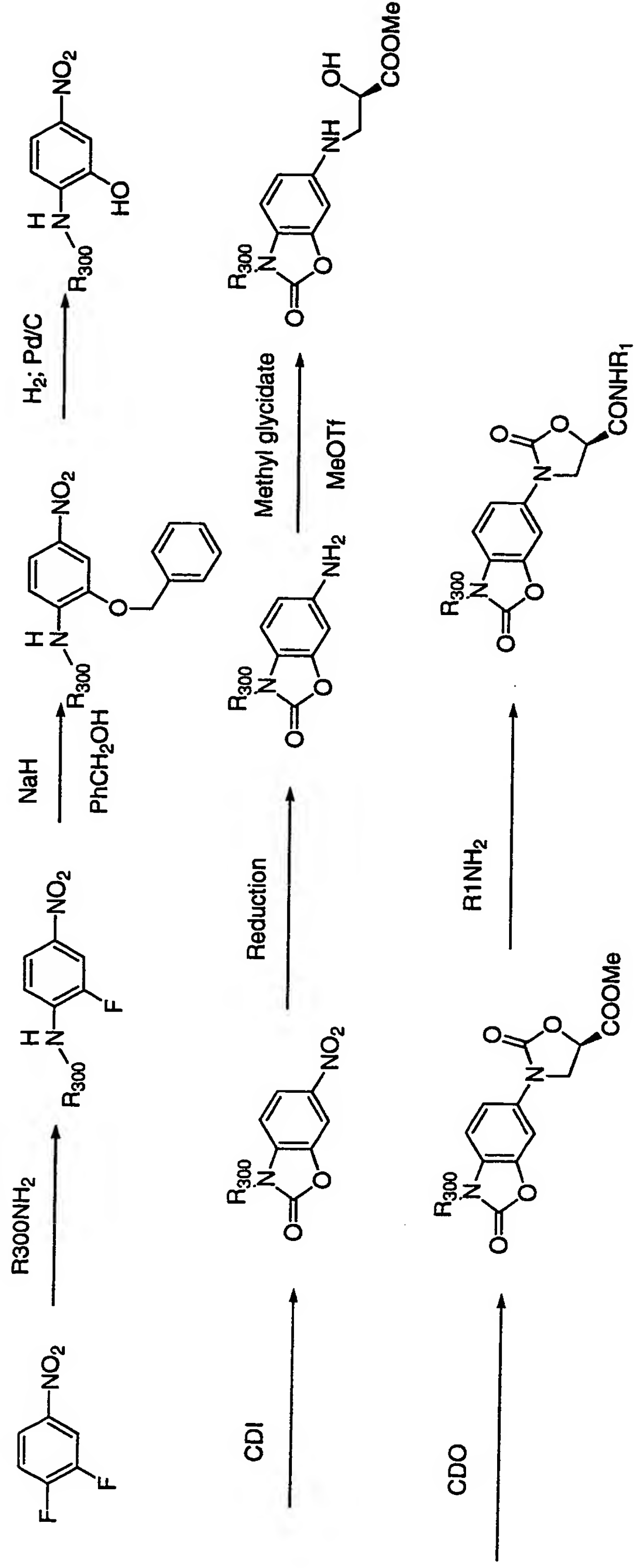


FIGURE XXVII

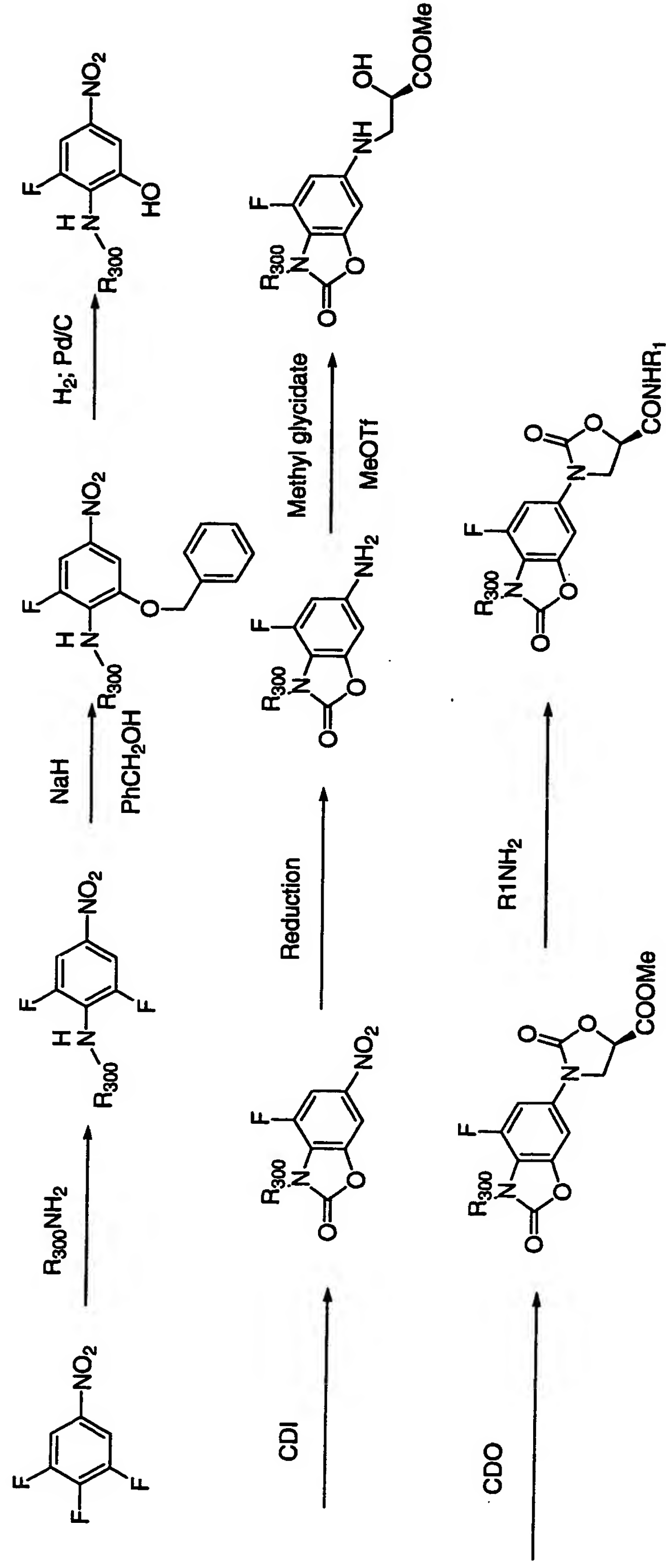
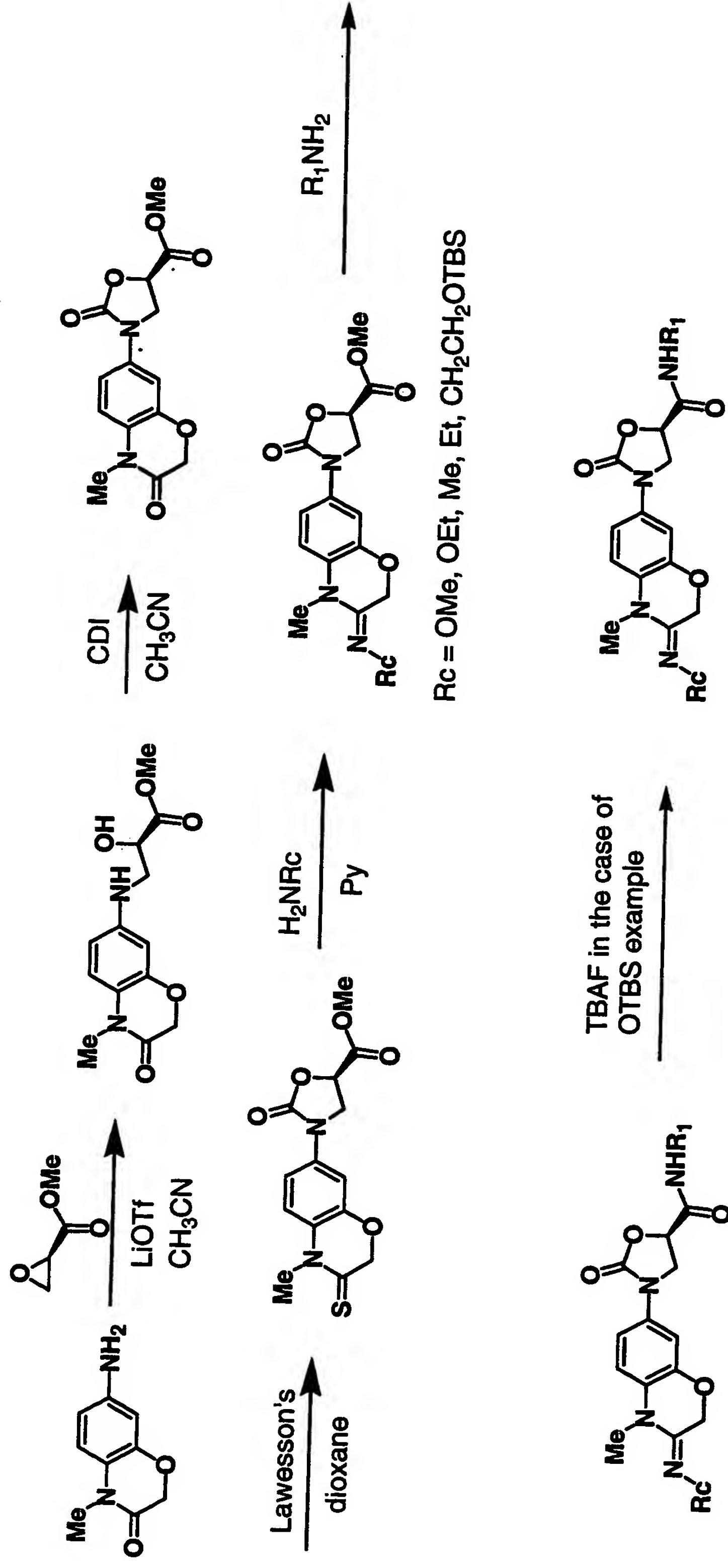


FIGURE XXVIII



Oc1cc(N)cc(F)c1[N+](=O)[O-]
 $\xrightarrow[100\text{ }^{\circ}\text{C}]{\text{NaOH, water}}$
O=C1OC(=O)Nc2cc(F)c([N+](=O)[O-])cc21

COCC(=O)Nc1cc(F)c([N+](=O)[O-])cc1
 $\xrightarrow[\text{DMF}]{\text{K}_2\text{CO}_3}$
COCC(=O)Nc1cc(F)c([N+](=O)[O-])cc1

COCC(=O)Nc1cc(F)c([N+](=O)[O-])cc1
 $\xrightarrow[\text{K}_2\text{CO}_3]{\text{MeI}}$
COCC(=O)Nc1cc(F)c([N+](=O)[O-])cc1

